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STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by shadow__7. Terms in this set (6) Explain how an ionic compound made up of charged particles can be electrically neutral. The total positive charge of the cations in the compound equals the total negative charge of the anions in the ...

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320 Chapter 10 • The Mole. Matt Meadows. The mole The mole, abbreviated mol, is The answer is less than 10 mol, as predicted, and has the correct unit, moles. Section. 10.1 Assessment. Moles to mass Now suppose that while working in a chemistry lab, you need 3.00 mol of copper (Cu) for a.....

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answer. 157.5 g 106.5 g 51 g Yes. Mass
of reactants equals mass of products.

Section 3.2 Assessment page 79 10.

Classify each example as a physical
change or a chemical change. a.

crushing an aluminum can physical b.

recycling used aluminum cans to make

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new aluminum cans physical c.
aluminum combining with oxygen to
form aluminum oxide chemical 11.

Matter—Properties and ChangesMatter—Properties and Changes

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Section 6.1 Assessment page 181 1.
Describe the development of the modern periodic table. Include contributions made by Lavoisier,

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Newlands, Mendeleev, and Moseley. Lavoisier organized a list of the known elements of his day as four categories. Newlands was the first to organize the elements and show that properties repeated in a periodic way ...

The Periodic Table and Periodic Law
The Periodic Table and ...

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Section 2.1 Units and Measurements

- Define SI base units for time, length,

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mass, and temperature. mass: a measurement that reflects the amount of matter an object contains • Explain how adding a prefix changes a unit.

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